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Modelling of an Iron Ore Reduction Furnace Operated with Pure Hydrogen

A. Ranzani da Costa; D. Wagner; F. Patisson

*Institut Jean Lamour, CNRS, Nancy-Université
Nancy - France*



Steel Production

- **Steel production (1950) $\Rightarrow \approx 200$ million tons worldwide^[1]**
- **Steel production (2007) $\Rightarrow \approx 1.3$ billion tons worldwide^[1]**
- **Steel consumption (2015) $\Rightarrow \approx 1.8$ billion tons^[2]**
 - future growth: 3-5%^[1]
 - 8-10% growth in BRIC countries (Brazil, Russia, India, China) ^[1] \Rightarrow dynamic expansion of their economies and infrastructures.
- **Steel**
 - Environment friendly material
 - can be recycled indefinitely
 - no loss of quality.
 - 40% of steel produced is recycled from scrap
 - Use in durable products/infrastructure
 - not enough scrap available.



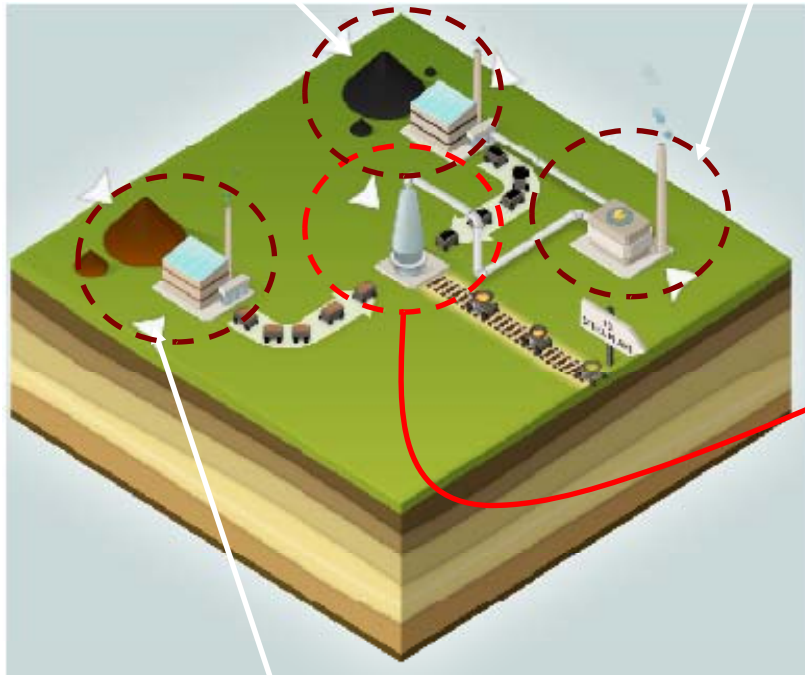
Ironmaking today – Blast Furnace (BF)

Coke Plant:

Coal → Coke

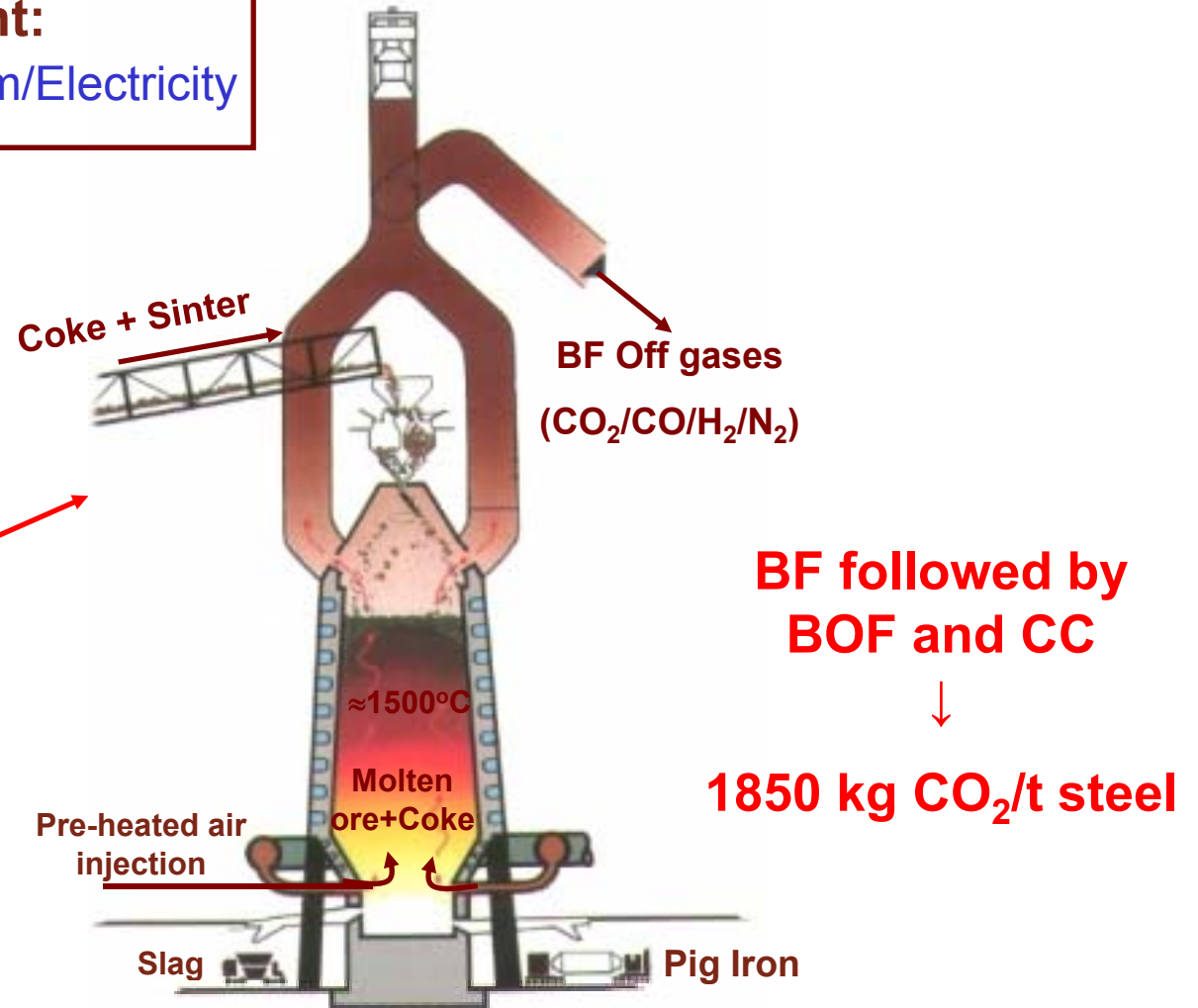
Power Plant:

BF Off gases → Steam/Electricity

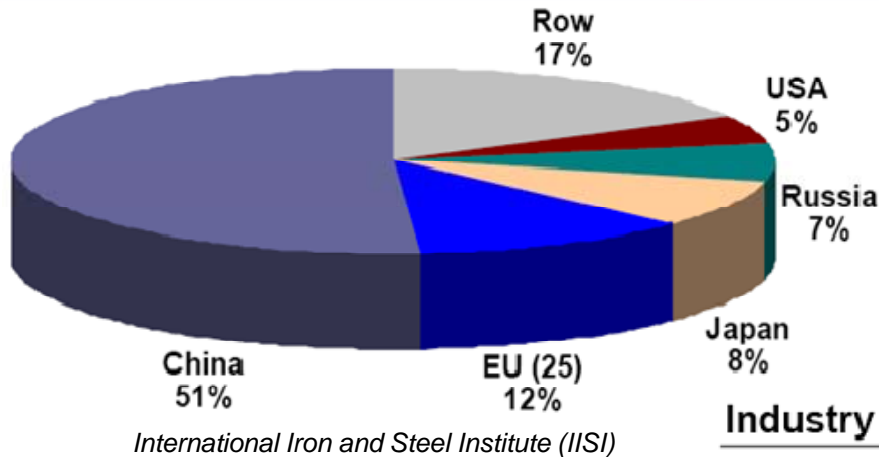


Pellet / Sinter Plant

Raw ore → heat treatments/additives
→ Sinter/pellets (better properties)



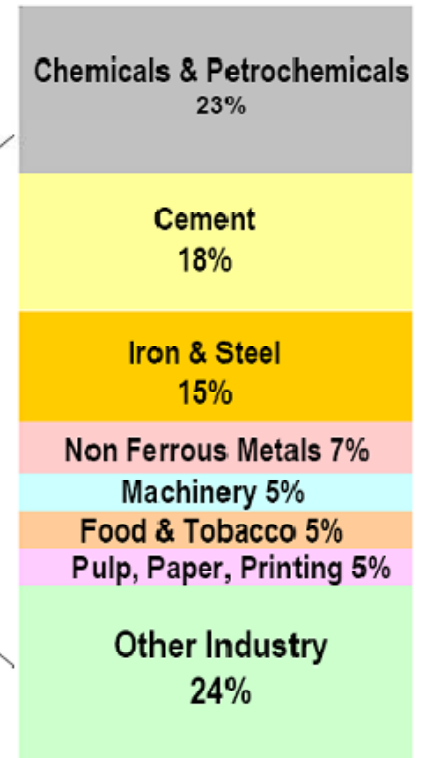
Steel: the Essential Numbers



International Iron and Steel Institute (IISI)

International Energy Agency (IEA)

Industry 21%



World steel industry today (2008)^[1]

- 1.3 billion tonnes of steel
- 1.9 ton CO₂ /ton steel
- 2.2 billion tonnes of CO₂
- 4 to 5% of total man-made greenhouse gases

CO₂ Emissions x Climate Change

- CO₂ Emissions ⇒ Climate Change ⇒ Natural disasters



- The best European steel plants are operating at the limits of what is presently technically possible.
- Breakthrough technologies for steelmaking industry

ULCOS (*Ultra Low CO₂* Steelmaking) Project

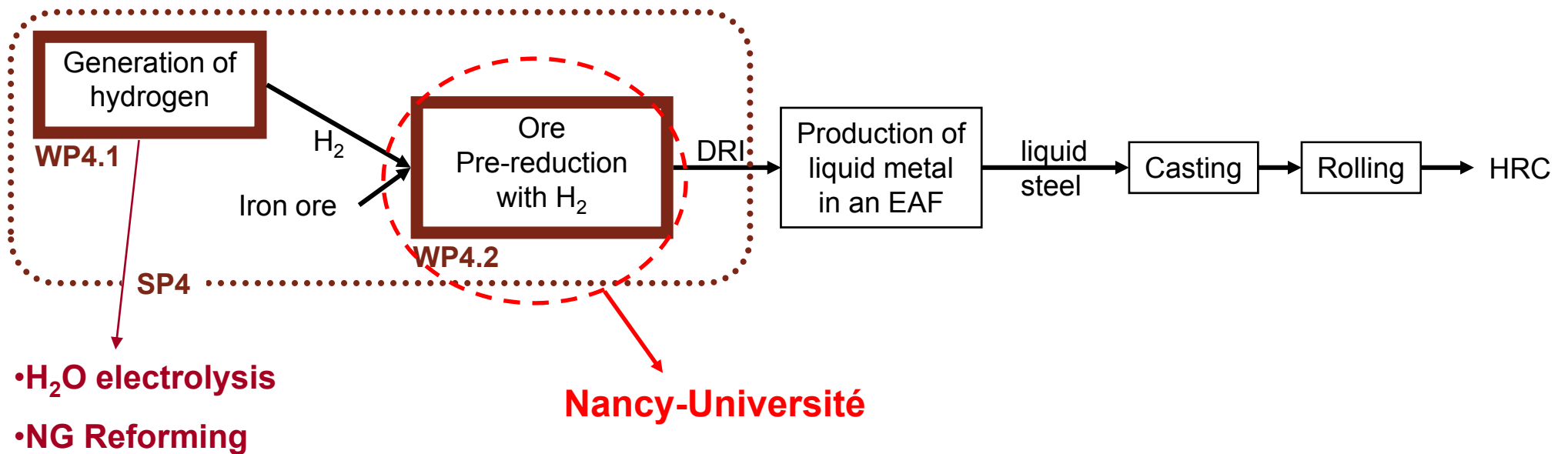
- Aim: design **new steelmaking processes able to reduce by more than 50% the CO₂ emissions** of the steel industry as compared to the current benchmark route (integrated plant, BF-CC)
- Launched by the major European steel companies at Sep/04
- Supported by European Commission
- 48 partners, 59 M€ budget, 5-year duration



ULCOS Breakthrough Technologies

- Optimization of the blast furnace + CCS
- New smelting technologies
- **Hydrogen steelmaking**
- Electrolysis + nuclear power

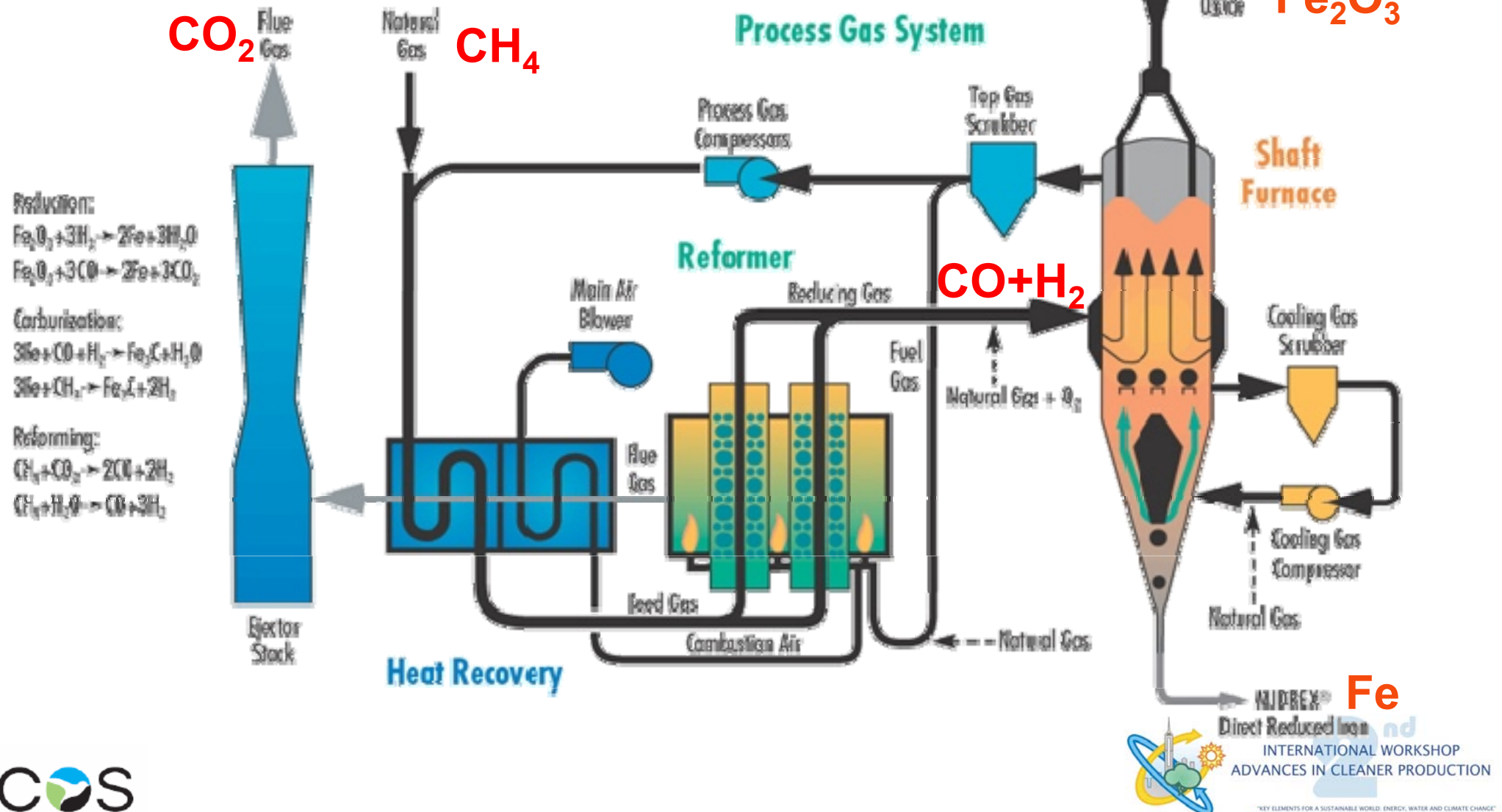
■ Hydrogen-based route defined



Low CO₂ Emissions: from 200 to 400 kg CO₂/t steel

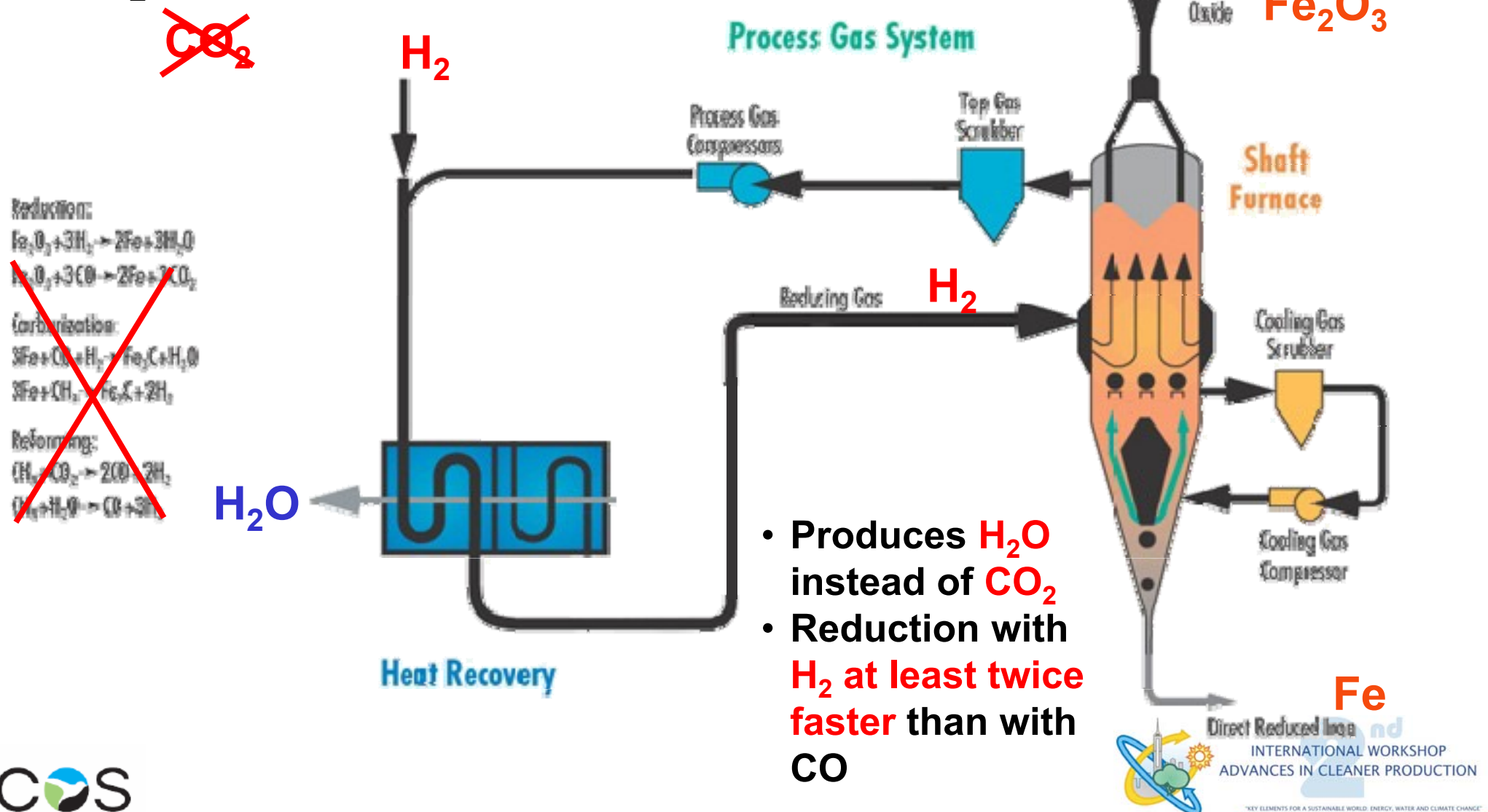
Using hydrogen for ironmaking

■ Usual NG-based direct reduction “MIDREX” process



Using hydrogen for ironmaking

■ Pure-H₂-based direct reduction process



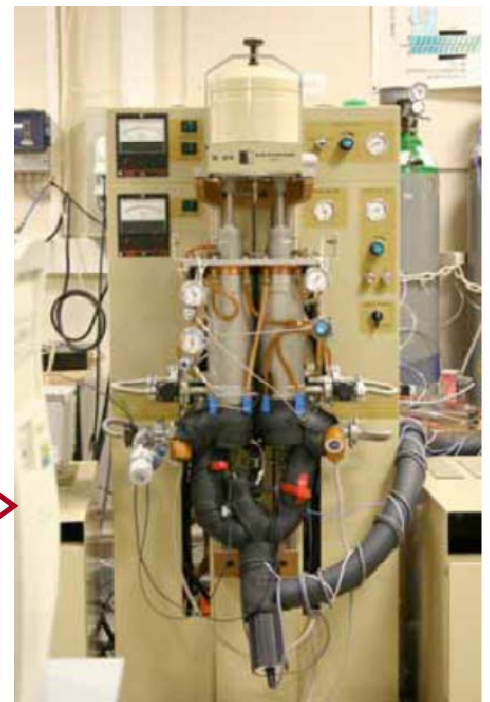
Let the hydrogen option open

- Major unknowns \Rightarrow future price of H_2 and the CO_2 cost of the H_2 generation (electricity).
- In the 2020's H_2 -economy may emerge, as a result of the evolution of transportation and energy industries.
- In this event, steelmakers should be ready to make massive use of H_2 .

Research at Nancy Université

■ Aim:

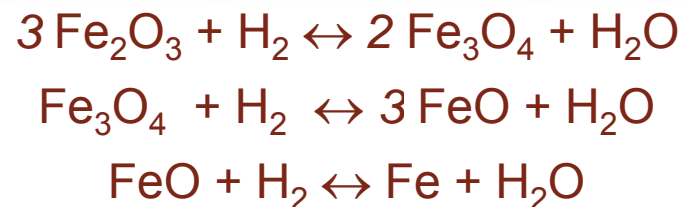
- Check the technical feasibility of hydrogen ironmaking and its performance from modelling work:
- Thermogravimetric experiments and sample characterizations to better understand the kinetic of the reactions
- Development of a pellet-scale kinetic model
- Development of a 2D numerical model of the reactor
- Study of the sticking phenomenon



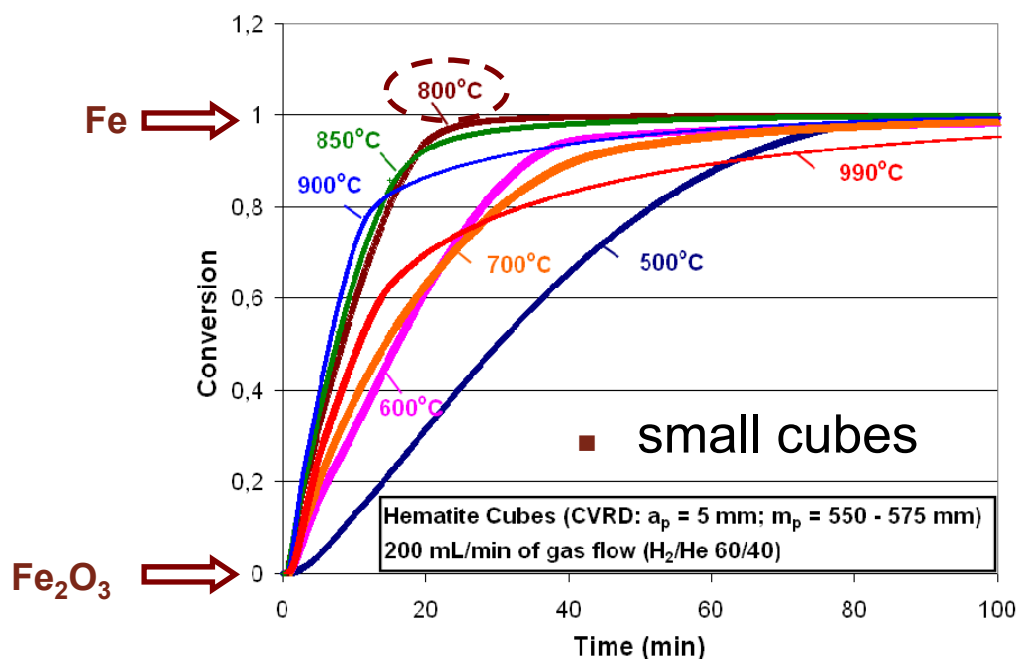
Main technical problems to be solved



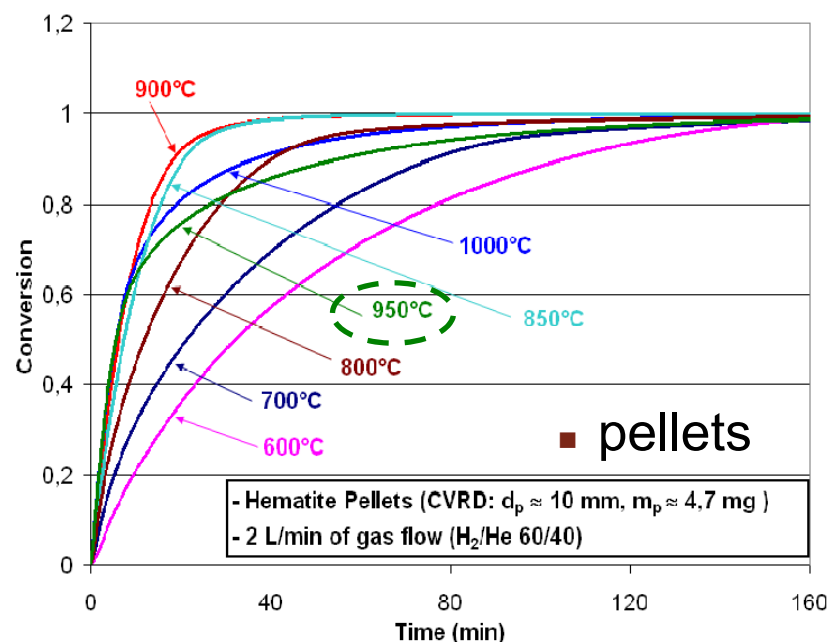
3 steps



Influence of the temperature on the reaction rate varies significantly with the raw material



Optimum at 800 °C

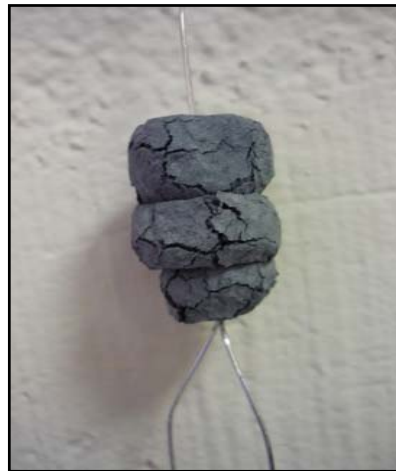
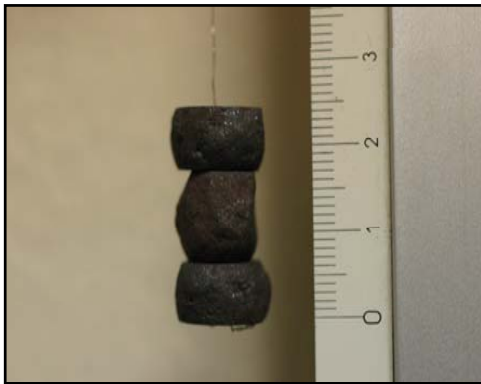


Minimum at 950 °C

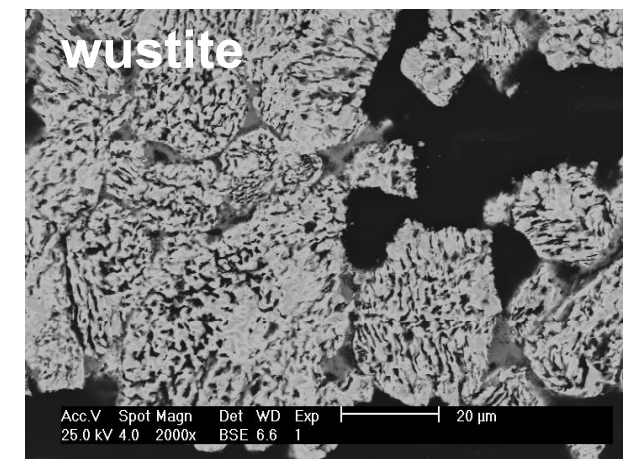
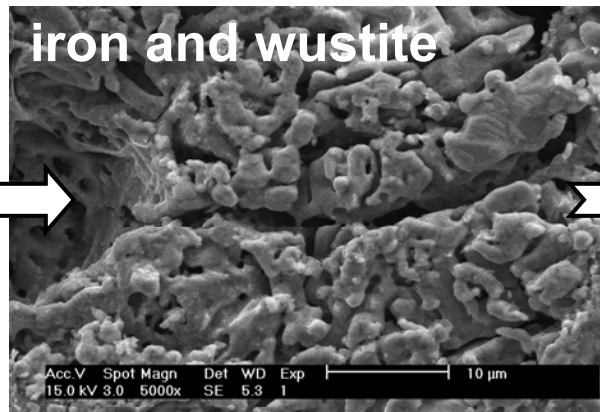
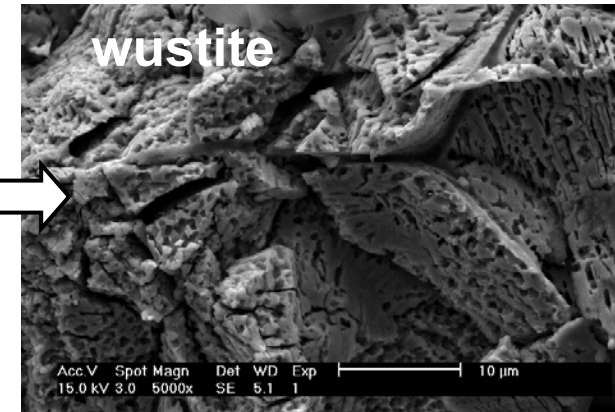
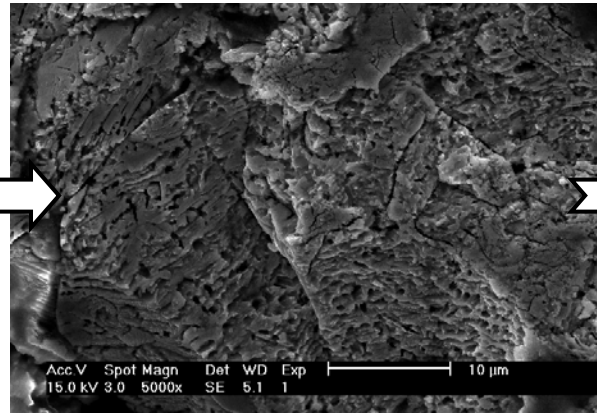
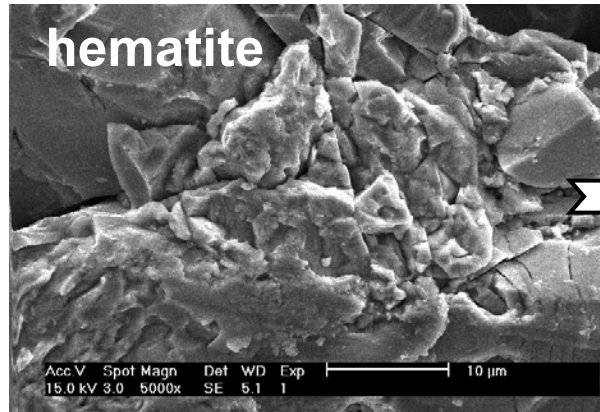
Main technical problems to be solved

Sticking Phenomenon

- H_2 reduction of DR grade pellets is faster at high temperatures.
- High temperatures seem to favour the “sticking”:
 - Reduced iron pellets stick to each other with a noticeable force.
 - Hinder or block the solid flow in shaft furnaces (MIDREX, HYL).
 - Lead to a complete defluidization of fluidized beds (FINMET).



Morphological changes

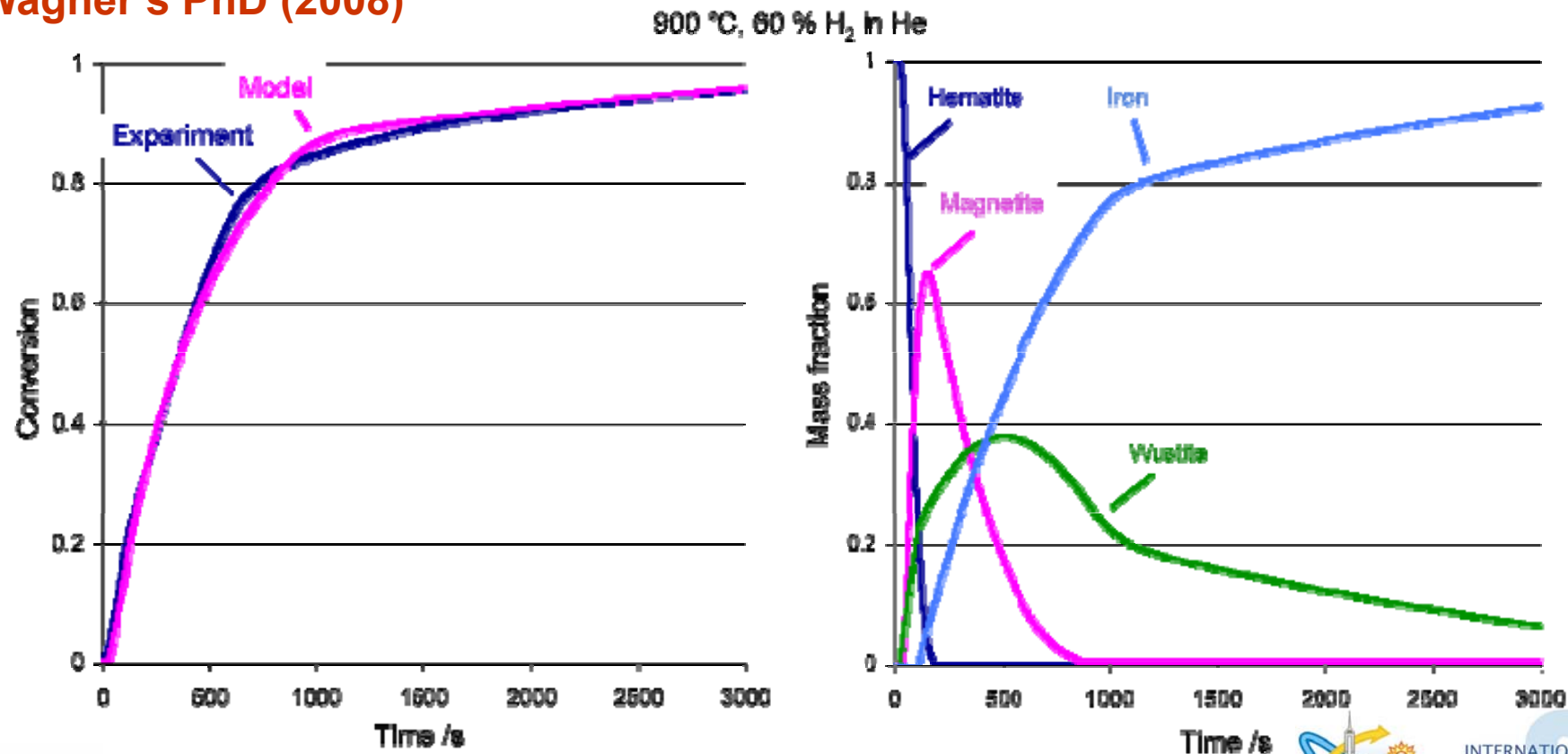


- Dense grains in hematite
- Grains become porous in magnetite and wustite
- Wustite grains break up into crystallites
- Molten-like structure of iron

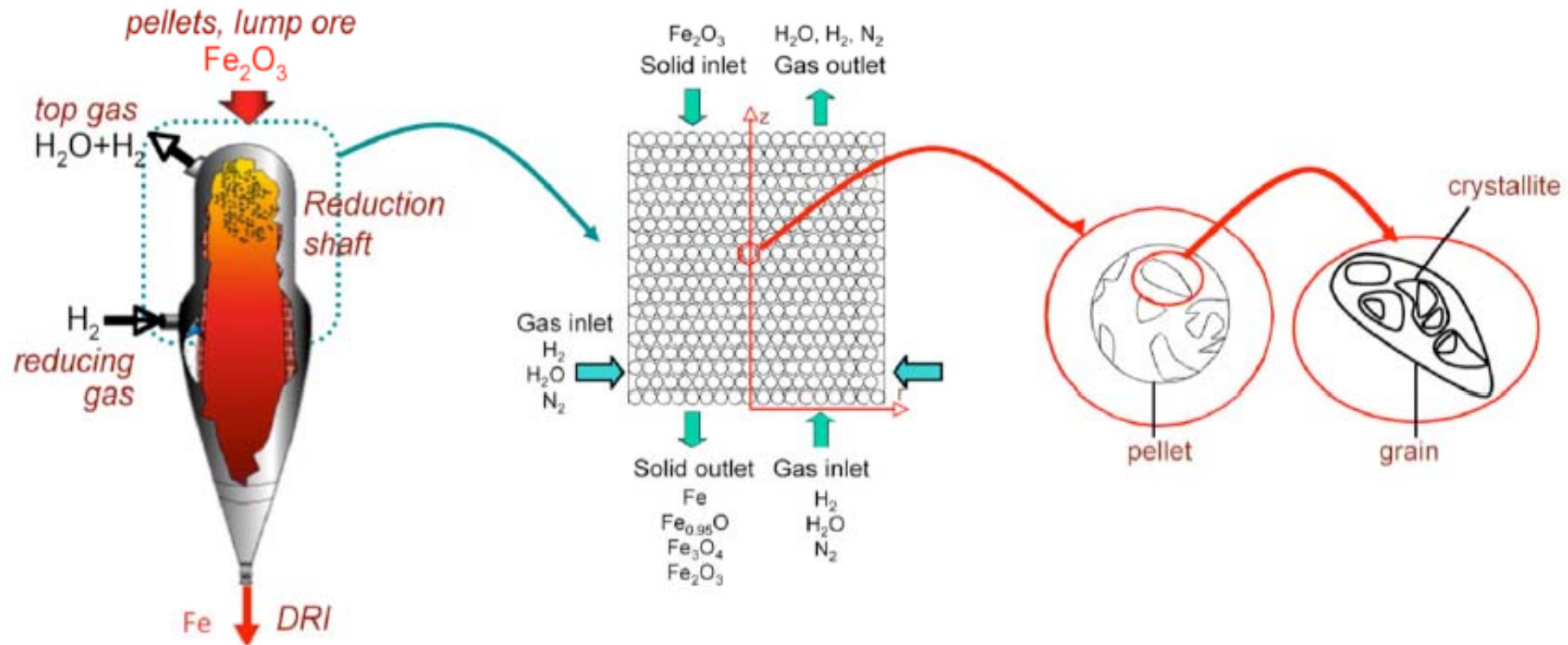
Kinetic model of a single pellet

- 3 reactions, mass transport (external transfer, inter- and intra-grain gas diffusion, solid diffusion, sintering of the iron layer)

D. Wagner's PhD (2008)



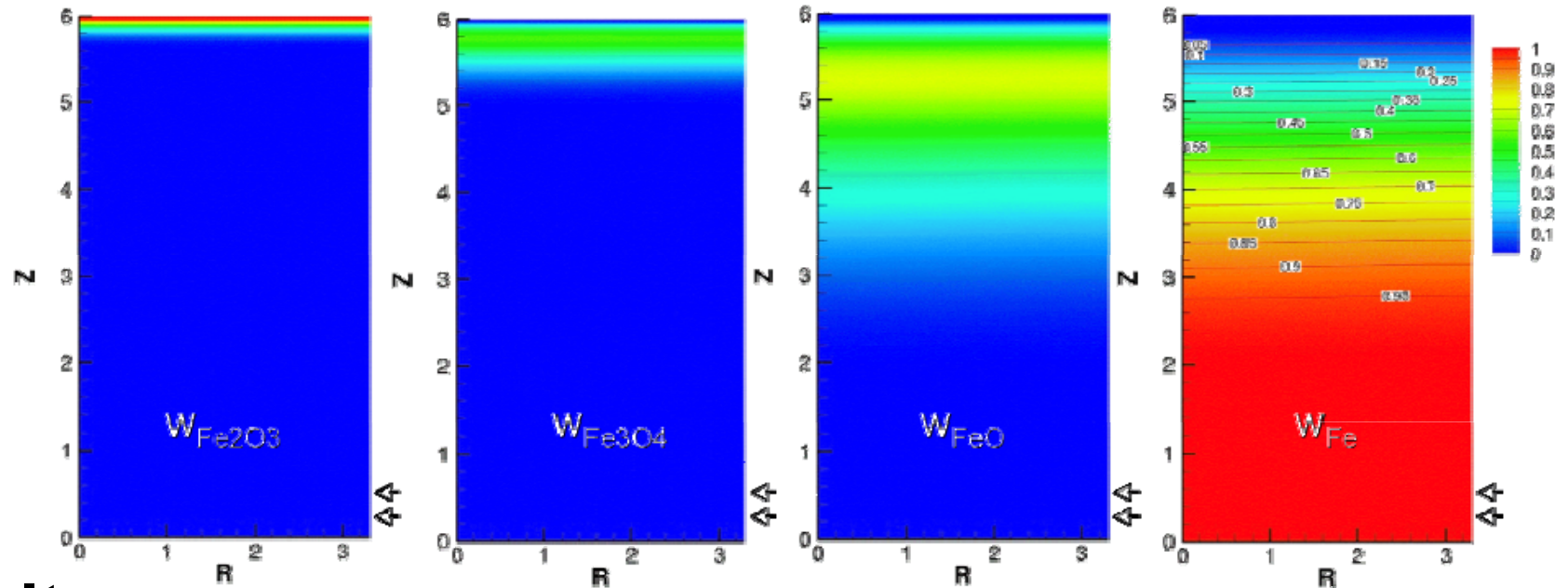
Numerical model of the H₂ shaft



- 2-D, axisymmetrical, steady-state model
- Written from scratch in FORTRAN 90 using the finite volume method
- Description of gas and solid flows, mass transfer, heat transfer, 3 reduction reactions
- Single-pellet sub-model included, with kinetics derived from thermogravimetry experiments

Numerical model of the H₂ shaft

D. Wagner's
PhD (2008)



■ Main results

- Reaction with H₂ is faster than that with CO
- Possible to use a more compact reactor than MIDREX and HYL shafts (e.g. 4.5 m instead of 9 m, with almost 100 % Fe⁰)
- Decreasing the pellet size clearly accelerates the process

Conclusions

- **Making steel using H_2 as a reducing gas makes sense:**
 - More than 50% reduction in CO_2 emission by several routes
 - Most of the technologies are mature
 - Social acceptability can be expected
 - Scientific problems are being studied
- **The main issues lie outside the steelmaking sector and expertise**
 - the future (2020, 2050) cost of hydrogen, which mostly depends on transport and energy policies
 - the CO_2 -cost of electricity.